

4-[(Z)-(Benzylamino)phenylmethylene]-5-methyl-2-phenyl-2H-pyrazol-3-one

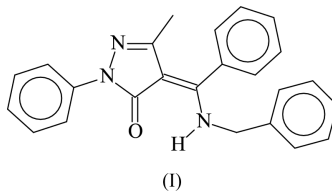
Ji-Jun Jiang,^a Xing-Qiang Lü,^a
Feng Bao,^a Bei-Sheng Kang^a and
Seik Weng Ng^{b*}^aSchool of Chemistry and Chemical Engineering, Sun Yat-Sen University, Guangzhou 510275, People's Republic of China, and ^bDepartment of Chemistry, University of Malaya, 50603 Kuala Lumpur, Malaysia

Correspondence e-mail: seikweng@um.edu.my

Key indicators

Single-crystal X-ray study
T = 293 K
Mean $\sigma(\text{C}-\text{C}) = 0.003 \text{ \AA}$
R factor = 0.043
wR factor = 0.134
Data-to-parameter ratio = 16.5For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.The NH unit on the exocyclic carbon-carbon double bond in the title compound, $\text{C}_{24}\text{H}_{21}\text{N}_3\text{O}$, lies on the same side of the double bond as the carbonyl unit of the pyrazolonyl ring, and the two interact through an N—H...O hydrogen bond [2.714 (2) Å].

Comment

This is a study on a compound that is related to 5-methyl-2-phenyl-4-[(Z)-(2-tolylamino)phenylmethylene]-2H-pyrazol-3-one (Bao *et al.*, 2004), one of a class of pyrazolones that are readily synthesized by condensing 1-phenyl-3-methyl-4-benzoyl-5-pyrazolone with a primary amine. The present compound (Fig. 1) has the isomeric benzylimino unit in place of the 2-tolylimino unit. No significant differences are found in the principal bond dimensions for the two compounds; the packing is similar, as noted from their calculated densities. The present compound also features an intramolecular hydrogen bond.

Experimental

1-Phenyl-3-methyl-4-benzoyl-5-pyrazolone (2.5 g, 9 mmol) and benzylamine (1 g, 9 mmol) were dissolved in ethanol (25 ml); the solution was heated under reflux for several hours. The solvent was removed and the pure product obtained upon recrystallization from ethanol in 75% yield.

Crystal data

 $\text{C}_{24}\text{H}_{21}\text{N}_3\text{O}$
 $M_r = 367.44$
Monoclinic, $P2_1/c$
 $a = 8.912 (1) \text{ \AA}$
 $b = 20.769 (3) \text{ \AA}$
 $c = 10.608 (1) \text{ \AA}$
 $\beta = 79.605 (2)^\circ$
 $V = 1931.2 (4) \text{ \AA}^3$
 $Z = 4$ $D_x = 1.264 \text{ Mg m}^{-3}$
Mo $K\alpha$ radiation
Cell parameters from 872 reflections
 $\theta = 2.5\text{--}25.9^\circ$
 $\mu = 0.08 \text{ mm}^{-1}$
 $T = 293 (2) \text{ K}$
Prism, yellow
 $0.50 \times 0.38 \times 0.14 \text{ mm}$

Data collection

Bruker SMART area-detector
diffractometer
 φ and ω scans
Absorption correction: none
11972 measured reflections
4250 independent reflections2456 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.032$
 $\theta_{\text{max}} = 27.2^\circ$
 $h = -11 \rightarrow 11$
 $k = -26 \rightarrow 23$
 $l = -10 \rightarrow 13$

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)] = 0.044$
 $wR(F^2) = 0.135$
 $S = 0.99$
 4250 reflections
 258 parameters
 H atoms treated by a mixture of independent and constrained refinement

$$w = 1/[\sigma^2(F_o^2) + (0.0631P)^2 + 0.2452P]$$

where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\max} = 0.001$
 $\Delta\rho_{\max} = 0.16 \text{ e } \text{Å}^{-3}$
 $\Delta\rho_{\min} = -0.15 \text{ e } \text{Å}^{-3}$

Table 1
 Selected geometric parameters (Å, °).

O1—C7	1.247 (2)	C9—C10	1.489 (3)
N1—C7	1.379 (2)	C11—C12	1.478 (2)
N1—N2	1.399 (2)	C18—C19	1.508 (2)
N1—C1	1.412 (2)	C19—C24	1.378 (2)
N2—C9	1.306 (2)	C19—C20	1.386 (2)
N3—C11	1.321 (2)	C20—C21	1.377 (2)
N3—C18	1.450 (2)	C21—C22	1.364 (3)
C7—C8	1.434 (3)	C22—C23	1.369 (3)
C8—C11	1.395 (2)	C23—C24	1.382 (2)
C8—C9	1.428 (2)		
C7—N1—N2	111.6 (1)	C9—C8—C7	105.4 (1)
C7—N1—C1	128.5 (2)	N2—C9—C8	111.7 (2)
N2—N1—C1	119.5 (1)	N2—C9—C10	118.9 (2)
C9—N2—N1	106.5 (1)	C8—C9—C10	129.4 (2)
C11—N3—C18	127.6 (2)	N3—C11—C8	119.2 (2)
C6—C1—N1	121.0 (2)	N3—C11—C12	118.2 (2)
C2—C1—N1	119.3 (2)	C8—C11—C12	122.6 (2)
O1—C7—N1	126.0 (2)	C13—C12—C11	121.3 (2)
O1—C7—C8	129.2 (2)	C17—C12—C11	119.1 (2)
N1—C7—C8	104.9 (2)	N3—C18—C19	114.4 (1)
C11—C8—C9	132.5 (2)	C24—C19—C18	123.5 (2)
C11—C8—C7	122.1 (2)	C20—C19—C18	117.7 (2)

H atoms were placed at calculated positions in the riding model approximation (C—H 0.93 Å for the aromatic H atoms, C—H 0.96 Å for the methyl H atoms and C—H 0.97 Å for the methylene H atoms), with $U_{\text{iso}} = 1.2U_{\text{eq}}$ (parent atom) for the aromatic and methyl C atoms, and $1.5U_{\text{eq}}$ for the methyl C atom. The amino H atom was located and refined.

Data collection: *SMART* (Bruker, 1999); cell refinement: *SAINT* (Bruker, 1999); data reduction: *SAINT*; program(s) used to solve

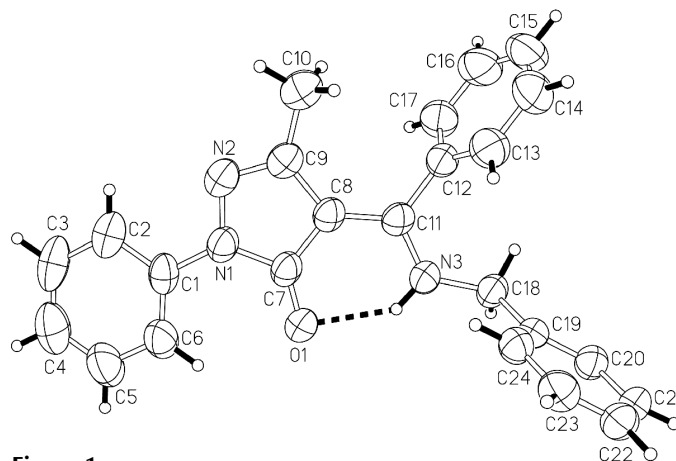


Figure 1
 ORTEPII (Johnson, 1976) plot of $C_{24}H_{21}N_3O$, with displacement ellipsoids drawn at the 50% probability level. H atoms are shown as spheres of arbitrary radii.

structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEPII* (Johnson, 1976); software used to prepare material for publication: *SHELXL97*.

The authors thank the National Natural Science Foundation of China, the Natural Science Foundation of Guangdong Province, Sun Yat-Sen University and the University of Malaya for supporting this work.

References

Bao, F., Lü, X.-Q., Wu, Q., Kang, B.-S. & Ng, S. W. (2004). *Acta Cryst.* **E60**, o155–o156.
 Bruker (1999). *SAINT* and *SMART*. Bruker AXS Inc., Madison, Wisconsin, USA.
 Johnson, C. K. (1976). *ORTEPII*. Report ORNL-5138. Oak Ridge National Laboratory, Oak Ridge, Tennessee, USA.
 Sheldrick, G. M. (1997). *SHELXS97* and *SHELXL97*. University of Göttingen, Germany.